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A Highly Stereoselective Formal Synthesis of (\pm) -Thienamycin through Organocopper Enolate–Iminoester Condensation

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The conjugate addition of the Fleming silylcuprate reagent to methyl crotonate and enolate trapping by methyl 4-methoxyphenyliminoacetate produced a high yield of $(1'S^*, 3R^*, 4S)$ -3-[1-(dimethylphenylsilyl)ethyl]-4-methoxycarbonyl-1-(4-methoxyphenyl)azetidin-2-one as a (\pm) -thienamycin building block.

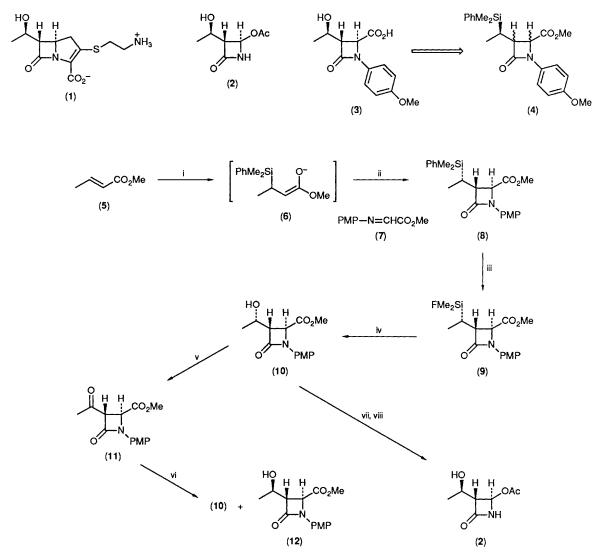
Thienamycin (1) and related β -lactam antibiotics comprise an interesting class of compounds, not only for their biological properties, but also for the chemistry involved in their total or formal synthesis.^{1,2} Most of the reported syntheses of (1) require, as a key step, the formation of a 3-(1-hydroxyethyl)-4-acetoxyazetidin-2-one (2), usually generated from a prefunctionalised monocyclic β -lactam, such as (3), and further elaboration at the C-4 position following the established Merck's methodology.³

Fleming and Kilburn⁴ reported a formal synthesis of (\pm) -thienamycin (1) by using the β -dimethylphenylsilyl group as a masked form of the hydroxy group, and recently we have reported⁵ that β -dimethylphenylsilylbutanoyl chlorideiminoester condensation could be an efficient alternative to the above approach leading to the corresponding β -lactam (4) in good yield, although with moderate stereoselectivity. Among other methods for the preparation of monocyclic β -lactams, the lithium enolate-imine condensation has received considerable attention in thienamycin synthesis.⁶ Following this approach, Hart *et al.*⁷ reported that the lithium enolate of ethyl β -(dimethylphenylsilyl)butyrate reacts with imines to afford β -lactams with modest to excellent stereoselectivity. Unfortunately, this approach fails when iminoesters are used as imino components of the reaction.† In general, the metal enolate–iminoester approach⁸ only works well with tin(II) enolates, silyl ketene acetals, and boron enolates,‡ to give the corresponding β -aminothiolesters or β -amino acids, which then were cyclised by standard procedures.¹⁰ In recent years, organocopper enolate chemistry¹¹ has found wide applicability in organic synthesis but, to our knowledge, no applications in β -lactam synthesis have been published.

We present here a concise formal synthesis of (\pm) -thienamycin (1) through an organocopper enolate-imine condensation, which involves a one-step preparation of a 3-(1dimethylphenylsilylethyl)- β -lactam of type (8), followed by further transformation of the dimethylphenylsilyl group into the required hydroxy functionality present in thienamycin. The key of the method is that addition of Fleming's

[†] Reaction of the lithium enolate of ethyl β -(dimethylphenylsilyl)butyrate with methyl 4-methoxyphenyliminoacetate did not lead to the desired β -lactam. See also ref. 16.

 $[\]ddagger$ For a recent study on the reaction of $\alpha\text{-iminoesters}$ with organometallic compounds, see ref. 9.



Scheme 1. Reagents and conditions: i, $(PhMe_2Si)_2CuCNLi_2$, tetrahydrofuran (THF), 0 °C, 20 min; ii, THF, 0 \rightarrow 20 °C, 3 h; iii, HBF₄·Et₂O, CH₂Cl₂, 0 °C \rightarrow room temp., 20 h; iv, MeCO₃H (32%), NEt₃, room temp., 3 h; v, NDC, CH₂Cl₂, room temp., 8 h, vi, NaBH₄, MeOH, 0 °C, 10 min; vii, PPh₃, diethyl azodicarboxylate (DEAD), HCO₂H, 0 \rightarrow 25 °C, 1.5 h, then MeOH, HCl, 25 °C, 1 h; viii, ref. 16. PMP = *p*-methoxyphenyl.

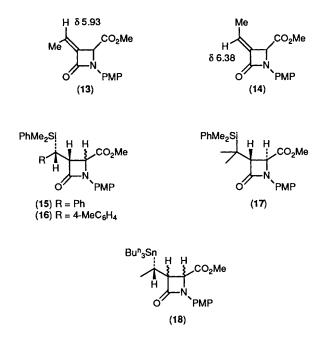
Table 1. Pre	eparation	of β-lactams	(15)-(17). ^a
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		¹ H NMR data ^d		
Compound ^b	Yield (%) ^c	trans : cis	trans $J_{1',3}$ $J_{3,4}$	$\begin{array}{c} cis \\ J_{1',3} J_{3,4} \end{array}$
(15)	77	85:15	10.5 2.4	13.5 5.7
(16)	80	92:8	10.2 2.4	13.4 5.5
(17)	85	100:0	- 2.5	

^a Reactions conducted on a 10 mmol scale by addition of the cuprate to the α , β -unsaturated ester at 0 °C. ^b All compounds are racemic and were characterized by their physical and analytical data. ^c Yields based on weight of product isolated by column chromatography. ^d Determined by 300 MHz NMR spectroscopy; *J* values in Hz.

silylcuprate reagent to methyl crotonate² (5), followed by trapping of the *in situ*-generated enolate (6) by the iminoester (7) (molar ratio 1.5:1) in tetrahydrofuran as solvent, afforded

the β -lactam (8) in 80% yield as the single diastereoisomer [δ 3.34 (dd, 3-H, J_{1',3} 5.1, J_{3,4} 2.4 Hz) and 4.16 (d, 4-H, J_{3,4} 2.4 Hz)]. Conversion of the β -lactam (8) into the (\pm)-thienamycin building block (10) was easily accomplished in two steps. Accordingly, reaction between (8) and $HBF_4 \cdot Et_2O$ followed by peracetic acid oxidation of the resulting fluoride (9) $(1'S^*, 3R^*, 4S^*)$ -3-(1-hydroxyethyl)-4-methoxyafforded carbonyl-1-(4-methoxyphenyl)azetidin-2-one (10) in 90% overall yield [δ 3.39 (dd, 3-H, $J_{1',3}$ 5.4, $J_{3,4}$ 2.4 Hz) and 4.43 (d, 4-H, $J_{3,4}$ 2.4 Hz)]. We assigned the relative stereochemistry between C-1' and C-3 in two ways. First, oxidation of (10) by 3-carboxypyridinium dichromate (NDC reagent)13 followed by sodium borohydride reduction of the resulting methylketone (11) afforded a mixture of the 1'-hydroxyethyl compound (10) and its epimer (12) in a 60 : 40 ratio. The ¹H NMR spectrum [δ 3.35 (dd, 3-H, $J_{1',3}$ 4.1, $J_{3,4}$ 2.7 Hz) and 4.60 (d, 4-H, $J_{3,4}$ 2.7 Hz)] of the 1'R*-isomer showed a smaller coupling constant for 1-H and 3-H.¹⁴ Alternatively, epimerisation at C-1 was successfully accomplished by the Mitsunobu reaction,15 using formic acid as the nucleophile, to afford the corresponding formate of (12), which, upon acid hydrolysis,¹⁶



furnished the desired hydroxy compound in nearly quantitative yield. Conversion of both epimers (10) and (12) into their corresponding methanesulphonates and further stereospecific elimination¹⁷ provided the respective Z-(13) and E-(14) alkenes, thus confirming unambiguously the above stereochemical assignment. Since both N-1 and C-4 groups in the β -lactam (12) can be easily elaborated to the β -lactam (2), our procedure constitutes a highly stereoselective formal synthesis of (±)-thieneamycin.

In order to determine the scope of the stereoselectivity of the reaction, we extended the method to other α , β -unsaturated esters and the results are summarized in Table 1. As can be seen, apart from the good yields obtained, the high degree of stereoselectivity seems to be the most important feature of the reaction. Further application of the present organocopper –enolate methodology is shown in the tributylstannylcuprate¹⁸ addition to methyl crotonate (**5**) and enolate trapping by the iminoester (**7**) to give the corresponding stannyl β -lactams (**18**) in 65% yield, as a mixture of *cis-trans*-isomers at the 3and 4-positions of the β -lactam ring in a 40:60 ratio.§

From the results reported here, \P the methodology developed could be readily extended to other synthetic applications in the β -lactam field and, consequently, in the chemistry that employs β -lactams as starting materials.

- § Although we have not assigned the relative configuration at C-3, the ¹H NMR spectra showed a $J_{1',3}$ 10.5 Hz and $J_{1',3}$ 3.9 Hz for the *trans*and *cis*-isomers respectively.
- \P All compounds prepared in this work are racemic, only one diastereoisomer is drawn.

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References

- 1 For reviews on β-lactam antibiotics, see: 'Chemistry and Biology of β-Lactam Antibiotics,' eds. R. B. Morin and M. Gorman, Academic Press, New York, 1982; vols. 1—3; 'Topics in Antibiotic Chemistry,' ed. P. G. Sammes, Ellis Horwood, New York, 1980, vols. 3 and 4; R. Southgate and S. Elson, in 'Progress in the Chemistry of Organic Natural Products,' eds. W. Herz, H. Grisebach, G. W. Kirby, and Ch. Tamm, Springer-Verlag, New York, 1985, p. 1; W. Durckheimer, J. Blumbach, R. Latrell, and K. M. Sheunemann, Angew. Chem., Int. Ed. Engl., 1985, 24, 980.
- 2 For reviews on carbapenem synthesis, see: T. Kametani, K. Fukumoto, and M. Ihara, *Heterocycles*, 1982, 17, 463; T. Nagahara and T. Kametani, *ibid.*, 1987, 25, 729; C. Palomo, in 'Recent Progress in the Chemical Synthesis of Antibiotics,' eds. G. Lukacs and M. Ohno, Springer-Verlag, Berlin, 1990, in the press.
- 3 P. J. Reider, R. Rayford, and E. J. J. Grabowski, *Tetrahedron Lett.*, 1982, 23, 379; P. J. Reider and E. J. J. Grabowski, *ibid.*, 1982, 23, 2293.
- 4 I. Fleming and J. D. Kilburn, J. Chem. Soc., Chem. Commun., 1986, 1198.
- 5 C. Palomo, J. M. Ontoria, J. M. Odriozola, J. M. Aizpurua, and I. Ganboa, J. Chem. Soc., Chem. Commun., 1990, 248.
- 6 For a recent review on thienamycin synthesis, see: G. I. Georg, in 'Studies in Natural Product Chemistry,' vol. 4, ed. Atta-ur-Rahman, Elsevier, Amsterdam, 1989, p. 431.
- 7 D. A. Burnett, J. C. Gallucci, and D. J. Hart, J. Org. Chem., 1985, 50, 5120.
- 8 For recent reviews on ester enolate-imine condensation, see: M. J. Brown, *Heterocycles*, 1989, **29**, 2225; D. J. Hart and D. C. Ha, *Chem. Rev.*, 1989, **89**, 1447.
- 9 Y. Yamamoto and W. Ito, Tetrahedron, 1988, 44, 5415.
- 10 For leading references, see: T. Yomada, H. Suzuki, and T. Mukaiyama, *Chem. Lett.*, 1987, 293; C. Gennari, G. Shimperna, and I. Venturini, *Tetrahedron*, 1988, 44, 4221; N. Miyachi, F. Kanda, and M. Shibasaki, *J. Org. Chem.*, 1989, 54, 3511.
- For recent reviews, see: R. J. K. Taylor, *Synthesis*, 1985, 364;
 B. H. Lipshutz, *ibid.*, 1987, 325.
- 12 I. Fleming, J. H. M. Hill, D. Parker, and D. Waterson, J. Chem. Soc., Chem. Commun., 1985, 318; for a review, see: I. Fleming, Pure Appl. Chem., 1988, 60, 71.
- 13 F. P. Cossio, M. C. López, and C. Palomo, *Tetrahedron*, 1987, 43, 3963.
- 14 G. Cainelli, M. Panuncio, T. Basile, A. Bongini, O. Giacomini, and G. Martelli, J. Chem. Soc., Perkin Trans. 1, 1987, 1637.
- 15 O. Mitsunobu, Synthesis, 1981, 1; for application in β-lactam chemistry, see: D. F. Corbett, S. Coulton, and R. Southgate, J. Chem. Soc., Perkin Trans. 1, 1982, 3011; D. G. Melillo, I. Shinkai, T. Liu, K. Ryan, and M. Sletzinger, Tetrahedron Lett., 1980, 21, 2783.
- 16 G. I. Georg, J. Kant, and H. S. Gill, J. Am. Chem. Soc., 1987, 109, 1129.
- 17 F. Pecquet and J. d'Angelo, Tetrahedron Lett., 1982, 23, 2777.
- 18 For a recent method for preparation of stannylcuprates, see A. C. Oehlaschlager, M. W. Hutzinger, R. Aksela, S. Sharma, and S. M. Sing, *Tetrahedron Lett.*, 1990, **31**, 165.